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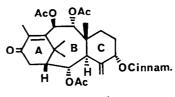
A SYNTHETIC APPROACH TO TAXANE DITERPENES. A SYNTHESIS OF THE BICYCLO[5.3.1]UNDECENONE RING SYSTEM

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Summary: Bicyclo[5.3.1]undecenone 2 corresponding to A and B rings in taxane diterpenes was synthesized. The eight-membered ring was constructed by a base-induced intramolecular cyclization of twelve-membered lactam sulfoxides **15**.

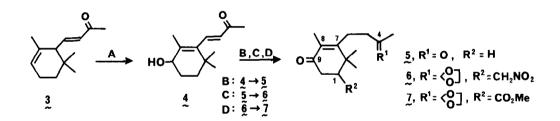
The crucial problem in the synthesis of the taxane diterpenes¹ such as taxinine (1) possessing unique tricyclic carbon skeleton is obviously in the construction of the sterically congested eight-membered B ring and thus several approaches involving fragmentation^{2a}, rearrangement^{2b} or intramolecular Diels-Alder reaction^{2c} have already been reported. Independently, we focused our attention on a direct closure of the eight-membered ring from a

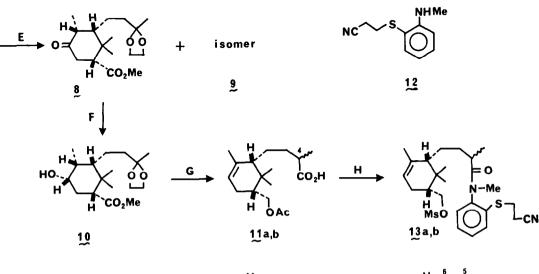
suitable precursor, by the novel intramolecular cyclization process we developed recently for the preparation of medium-ring ketones.³ We report here a synthesis of bicyclo[5.3.1]undec-8-en-3-one 2 corresponding to taxane A and B rings and having functionality in the B ring for the construction of C ring.

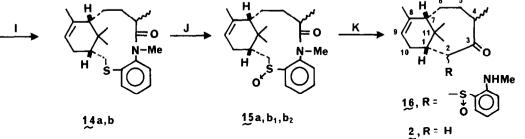


Taxinine (1)

Selective epoxidation of the double bond in cyclohexane moiety of α -ionone (3) followed by K₂CO₃-MeOH treatment afforded dienyl ketone 4. The double bond next to the carbonyl group was then selectively reduced with NaTeH⁴ and the resulting keto allyl alcohol was oxidized with CrO₃ to give diketone 5 in 62 % yield.^{5,7} Conversion of 5 into 6 having a nitromethyl group at C-1 position was accomplished in three steps: i) preferential acetalization of saturated ketone in the presence of α , β -unsaturated ketone by ethylene glycol-TsOH, ii) DDQ oxidation to give cross-conjugated dienone, iii) Michael addition of CH₃NO₂ to C-1 position. Conversion of a nitromethyl group into a methoxycarbonyl group was effected by TiCl₃-MeONa treatment,⁸ Jones' oxidation, methylation with dimethylsulfate and reacetalization. High-pressure catalytic hydrogenation of 7 and the subsequent PCC oxidation of a C-9 hydroxyl group afforded 8 (86 %)⁹ along with a small amount of isomer 9 (10 %). Keto ester 8 was found to be a desired compound having cis-diequatorial substituents at C-1 and C-7 positions. The predominant formation of 8 over 9 may be considered as follows:

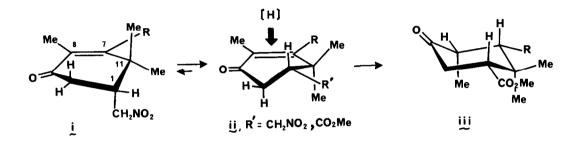






(A) mCPBA/CH₂Cl₂, K₂CO₃/MeOH (reflux), (B) NaTeH/EtOH, Jones' reagent (62 %, $3 \rightarrow 5$, 4 steps), (C) ethylene glycol/TsOH/benzene, DDQ/benzene (80°C) (69.5 %, 2 steps), CH₃NO₂/i-Pr₂NH/DMSO (75°C, 97 % based on consumed material), (D) MeONa/TiCl₃/AcONH₄/ aq.MeOH, Jones' reagent, Me₂SO₄/K₂CO₃, ethylene glycol/TsOH/benzene (67.5 %, 4 steps); (E) H₂(100 atm)/Pd-C/AcOEt, PCC/AcONa/CH₂Cl₂ (92 %); (F) NaBH₄/EtOH (-10°C, 93 %,); (G) Ph₃P/(NCO₂Et)₂/THF (78 %), LiAlH₄/Et₂O, Ac₂O/Py, TsOH/acetone (95 %, 3 steps), Ph₃P=CHOMe/THF, Ac₂O/Py, HClO₄ aq., Jones' reagent (70 %, 4 steps); (H) 1N KOH, MsCl/Et₃N/CH₂Cl₂, (COCl)₂/benzene, 12/K₂CO₃/Et₂O (46 %, 4 steps), (I) K₂CO₃/NaBH₄/DMF (135°C, 70 %), (J) NaIO₄/aq.MeOH; (K) LDA/THF (-65° – 0°C, 60-65 % based on consumed material), Na-Hg/Na₂HPO₄/MeOH (86 %, a/b ratio, 5/3).

Initial Michael addition of CH_3NO_2 to the cross-conjugated dienone derivative of 5 may proceed from the direction perpendicular to the π -system affording i. However, ring inversion should take place giving the more stable conformer ii, in which the nitromethyl group is in quassiequatorial position. The methoxycarbonyl derivative should hold the same conformation ii as the nitromethyl derivative.¹⁰ In conformer ii (R'= CO_2Me), the α -side is sterically much more hindered than the β -side due to a quassi-axial methyl group at C-11 position and thus catalyst may approach from the β -side giving mainly 1,7-cis (diequatorial) **8** (iii).



Sodium borohydride reduction of 8 produced the C-9 α -alcohol 10 predominantly (α/β ratio, 40:1). Alcohol **10** was treated with Ph₃P and (NCO₂Et)₂ to give Δ^8 -unsaturated ester,¹¹ which was converted into an epimeric mixture (1:1) of **11a,b** at C-4 by a series of reactions (LiAlH_{Δ} reduction, acetylation, hydrolysis to give ketone, one-carbon elongation with $Ph_3P=CHOMe_*$ reacetylation, acidic hydrolysis to give aldehyde and Jones' oxidation). Compounds 11 were, after conversion of the acetate into the mesylate and a carboxyl group into the acid chloride, treated with amino sulfide **12^{3b}** yielding a mixture (1:1) of amide **13a,b.** Removal of a S-protecting group from **13a,b** and the concomitant twelve-membered lactam sulfide formation was effected with K_2CO_3 in DMF in the presence of NaBH $_4$ at 135°C giving 14a,b as an epimeric mixture at C-4 (70 %, a/b ratio, 2:3).¹² On NaIO₄ oxidation, 14a afforded sulfoxide 15a and 14b produced a mixture of $15b_1$ and $15b_2$.¹³ Each of the isomers was subjected to a LDAinduced intramolecular cyclization and a mixture of eight-membered keto sulfoxides 16 was obtained in 60-65 % yield (based on the consumed starting material) in every case. Mixtures of 16 obtained from different sources, 15a, b1, b2, were combined and subjected to desulfurization with Na-Hg in MeOH in the presence of Na_2HPO_4 producing the desired bicyclic ketones **2a,b** as a separable mixture of stereoisomers at C-4 (86 %, **a/b** ratio, 5:3).¹⁴

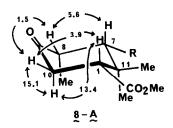
The present strategy for the formation of eight-membered B part of taxane skeleton would be applicable in general for the synthesis of more complex analogues. Synthesis of taxane ring skeleton from a derivative of **2** will be published in the forthcoming paper from this laboratory.

References and Notes

1. Several compounds belonging to <u>Taxus</u> species are reported to have potent antileukemic and tumor inhibitory properties: M.C. Wani, H.L. Taylor, M.E. Wall, P. Coggon and A.T.

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- 2. a) B.M. Trost and H. Hiemstra, <u>J. Am. Chem. Soc.</u>, **104**, 886 (1982); R.A. Holton, <u>ibid.</u>, **106**, 5731 (1984); H. Nagaoka, K. Ohsawa, T. Takata and Y. Yamada, <u>Tetrahedron Lett.</u>, **25**, 5389 (1984); T. Kojima, Y. Inouye and H. Kakisawa, <u>Chem. Lett.</u>, 323 (1985); b) S.F. Martin, J.B. White and R. Wagner, <u>J. Org. Chem.</u>, **47**, 3190 (1982); c) K.J. Shea and P.D. Davis, <u>Angew. Chem. Int. Ed. Engl.</u>, **22**, 419 (1983); K. Sakan and B.M. Craven, <u>J. Am. Chem. Soc.</u>, **105**, 3732 (1983); P.A. Brown, P.R. Jenkins, J. Fawcett and D.R. Russell, <u>J. Chem. Soc. Chem. Commun.</u>, 253 (1984); <u>cf</u>.) A.S. Kende, M. Benechie, D.P. Curran and P. Fludzinski, <u>Tetrahedron Lett.</u>, 4513 (1979); Y. Inouye, C. Fukaya and H. Kakisawa, <u>Bull. Chem. Soc. Jpn.</u>, **54**, 1117 (1981); R.Z. Andriamialisoa, M. Fetizon and I. Hanna, Tetrahedron, **40**, 4285 (1984).
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- 4. M. Yamashita, Y. Kato and R. Suemitsu, Chem. Lett., 847 (1980).
- 5. Diketone 5 has already been prepared from 3 by epoxidation followed by MeONa treatment, but the yield of 5 was only 8.3 $\%.^6$
- 6. M. Rosenberger, P. McDougal and J. Bahr, J. Org. Chem., 47, 2130 (1982).
- 7. All new compounds have been satisfactorily characterized by IR, ¹H NMR (60 or 400 MHz) and high resolution mass spectrometry (or elemental analysis).
- 8. J.E. McMurry and J. Melton, J. Org. Chem., 38, 4367 (1973).
- 9. The stereostructure of **8** was deduced as **8-A** by ¹H NMR (400 MHz) analysis. This assignment was confirmed by conversion into **2** as described later. ¹H-NMR(CDCl₃, 400 MHz) **8**: δ 1.09, 1.12, 1.40 and 3.70 (each s, Me), 1.14 (d, J= 7.6 Hz, 8 α -Me), 2.30 (10 β -H), 2.57 (8 β -H), 2.61 (1 β -H), 2.96 (10 α -H), 3.9-4.0 (m, 4H). By-product **9** is considered to be a stereoisomer of **8** from ¹H NMR data, but its stereostructure is remained unknown.



10. Even if epimerization takes place during conversion of **6** into **7**, the resulting **ii** having a quassi-ax CO_2Me group may be flipped into a more stable **i** having a quassi-eq CO_2Me group, which is an enantiomer of **ii** having a quassi-eq CO_2Me group.

- 11. The corresponding Δ^9 -isomer was not detected in this reaction.
- 12. ¹H NMR (CDCl₃) 14a(less polar gum): δ 0.89, 1.36 and 3.23 (each s, Me), 1.13 (d, J= 6.7 Hz, Me), 1.54 (d, J= 1.7 Hz, Me). 14b(more polar gum): δ 0.79 and 0.92 (each s, Me), 1.08 (d, J= 6.5 Hz, Me), 1.65 (d, J= 1.5 Hz, Me), 3.15 (s, 2.5H), 3.42 (s, 0.5H).
- 13. Compounds b₁, b₂ are stereoisomers due to sulfoxide moiety. ¹H NMR(CDCl₃) 15a(86 %): δ 0.99, 1.23 and 3.37 (each s, Me), 1.16 (d, J= 6.7 Hz, Me), 1.55 (d, J= 1.9 Hz, Me). 15b₁(less polar, 61 %): δ 0.86, 1.02 and 3.15 (each s, Me), 1.07 (d, J= 6.5 Hz, Me), 1.62 (d, J= 1.4 Hz, Me). 15b₂(more polar, 27 %): δ 1.00, 1.51 and 3.40 (each s, Me), 1.19 (d, J= 6.7 Hz, Me), 1.68 (d, J= 1.2 Hz, Me).
- 14. **2a** (less polar oil, 53.5 %): ¹H NMR(CDCl₃) δ 0.96 (d, J= 6.8 Hz, 4-Me), 0.98 and 1.18 (each s, 11 α and 11 β -Me), 1.62 (d, J= 1.6 Hz, 8-Me), 5.25- 5.55 (m, 9-H), IR(CCl₄) 1720(sh), 1695 cm⁻¹, High resolution MS(m/z) Calcd for C₁₅H₂₄O(M⁺): 220.1826, Found: 220.1837. **2b** (more polar oil, 32 %): ¹H NMR(CDCl₃) δ 0.97 and 1.15 (each s, 11 α and 11 β -Me), 1.04 (d, J= 7 Hz, 4-Me), 1.64 (d, J= 1.7 Hz, 8-Me), 5.15- 5.45 (m, 9-H), IR(CCl₄) 1735(sh), 1685 cm⁻¹, High resolution MS(m/z) Found: 220.1820.

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